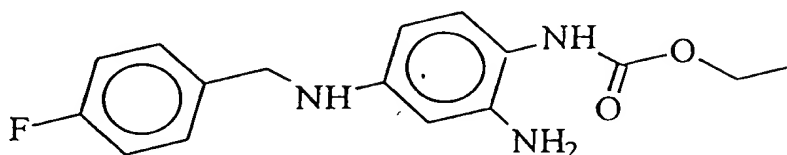


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~~Patent Claims~~

WHAT IS CLAIMED IS;

1. Modification A of the compound I



characterized by the X-ray diffractogram, reflections not coinciding with the reflections of the other two modifications being observed, inter alia, at  $6.97^{\circ}2\theta$  ( $12.67 \text{ \AA}$ ),  $18.02^{\circ}2\theta$  ( $4.92 \text{ \AA}$ ) and  $19.94^{\circ}2\theta$  ( $4.45 \text{ \AA}$ ).

2. Modification B of the compound I characterized by the X-ray diffractogram, reflections not coinciding with the reflections of the other two modifications being observed, inter alia, at  $15.00^{\circ}2\theta$  ( $5.90 \text{ \AA}$ ),  $19.29^{\circ}2\theta$  ( $4.60 \text{ \AA}$ ) and  $19.58^{\circ}2\theta$  ( $4.53 \text{ \AA}$ ).

3. Modification C of the compound I characterized by the X-ray diffractogram, reflections not coinciding with the reflections of the other two modifications being observed, inter alia, at  $9.70^{\circ}2\theta$  ( $9.11 \text{ \AA}$ ) and  $21.74^{\circ}2\theta$  [sic] ( $4.09 \text{ \AA}$ ).

4. Process for the preparation of the modification A according to Claim 1, characterized in that the pure crystal form is allowed to crystallize out of a supersaturated solution of the compound I in protic, dipolar-aprotic or non-polar solvents.

5. Process for the preparation of the modification A according to Claim 4, characterized in that the crystallization from the solution is carried out at

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temperatures from -20°C to 110°C, preferably at 20°C to 50°C.

6. Process for the preparation of the modification A according to Claims 4 and 5, characterized in that  
5 protic solvents which can be employed are lower alcohols such as ethanol, 2-propapanol [sic] or n-butanol, dipolar-aprotic solvents are acetonitrile or acetone and the non-polar solvent is toluene.

7. Process according to Claim 6, characterized in  
10 that lower alcohols are preferably used as solvents.

8. Process for the preparation of the modification A according to Claim 1, characterized in that the  
substance of the modifications B and C are [sic]  
15 treated with protic, dipolar-aprotic or non-polar solvents at low temperatures, preferably at room temperature.

9. Process for the preparation of the modification B according to Claim 2, characterized in that the pure crystal form is allowed to crystallize  
20 out at a temperature of greater than 80°C from a saturated solution of the compound I in protic or non-polar solvents.

10. Process for the preparation of the modification B according to Claim 9, characterized in that the  
25 protic solvent preferably employed is water and the non-polar solvent is toluene.

11. Process for the preparation of modification B according to Claim 2, characterized in that the  
modification B is preferably prepared from the  
30 modification A at temperatures of greater than 80°C by thermal phase conversion.

12. Process for the preparation of the modification C according to Claim 3, characterized in that the pure crystal form is preferably allowed to  
35 crystallize out at a temperature of from 50°C to 70°C from a saturated solution of the compound I in protic or alternatively non-polar solvents.

13. Process for the preparation of the modification C according to Claim 12, characterized in

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that the protic solvents employed is [sic] preferably ethanol and 2-propanol and the non-polar solvent is toluene.

14. Process for the preparation of the  
5 modification C according to Claim 12, characterized in that the crystallization from the solution is preferably carried out at temperatures from 60°C to 70°C.

15.<sup>4</sup> Use of the modification A, B and [sic] C of the  
10 compound I for the production of pharmaceutical preparations.

~~16.~~<sup>4</sup> Pharmaceuticals comprising the modification A, B or C of the compound I and, if appropriate, exipients and/or auxiliaries.

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